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A New Route to 3,4-Dihydro-2*H*-1-Benzopyrans Substituted at 3-Position *Via* Palladium-Catalysed Reactions

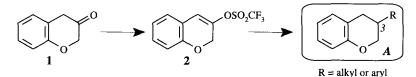
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Abstract: 3,4-Dihydro-2*H*-1-benzopyrans substituted at 3-position were prepared *via* palladium-catalysed reactions between a triflate and several coupling reagents (alkyl or aryl tin reagents and borane derivatives) according to Stille or Suzuki methodologies. © 1997 Published by Elsevier Science Ltd.

As far as we know, only two examples corresponding to the formula A were described in the literature (Scheme 1). On the one hand, Burali *et al*¹ prepared 3-phenyl-3,4-dihydro-2*H*-1-benzopyran in four steps and moderate overall yield from 2-hydroxybenzylalcohol. On the other hand, Beckwith *et al*² reported the synthesis of 3-methyl-3,4-dihydro-2*H*-1-benzopyrans *via* a radical reaction with the corresponding bromoarene. In this way the authors obtained a mixture of benzopyran and benzofuran derivatives with modest yields. Moreover, with this methodology the benzofurans were isolated as major products.

Scheme 1



Herein, we report the preparation of 3,4-dihydro-2*H*-1-benzopyrans substituted at the 3-position by different alkyl or aryl groups using chromanone 1^3 as starting material (Scheme 1). Numerous attempts to introduce methyl or ethyl groups *via* a Wittig reaction were unsuccessful, leading predominantly to degradation products. Consequently, we decided to run a palladium-catalysed reaction between triflate 2, generated directly from the chromanone 1, and several coupling reagents owing to Stille and Suzuki methodologies (Scheme 2).^{4,5}

The triflate 2 was obtained in a satisfactory yield starting from the the chromanone 1 using triflic anhydride and diisopropylethylamine in methylene chloride.⁶ Scheme 2 reports the synthesis of compounds 3 - 12 prepared in two steps from the triflate 2. The first reaction entails a coupling procedure between compound 2 either with trialkyltin derivatives in hexamethylphosphoramide with bis(triphenylphosphine)palladium (II) chloride at 65°C, or with boric acid derivatives in a mixture of

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toluene/ethanol with tetrakis(triphenylphosphine)palladium (0) under reflux. The second reaction involved an hydrogenation of the unsaturated intermediates providing respectively the expected compounds 3 - 9 and 10 - 12.

2		Scheme 2 1. $(Ph_3P)_2PdCl_2$ or $(Ph_3P)_4Pd$, Coupling reagent, solvent, time 2. H_2 , Pd/C, MeOH/CH ₂ Cl ₂ , 18 h, R.T. 3 - 12		
Cpd ^a	R	Coupling reagent	Time (h)	Yield (%) ^b
3	$\langle \rangle$	benzyltributyltin ^c	3	70
4	\square_{0}	[·] 2-(tributylstannyl)furan	2	78 ⁴
5	\square_{s}	2-(tributylstannyl)thiophene	2	74
6	CH ₃	tetramethyltin	3	42
7	CH ₂ CH ₃	tributyl(vinyl)tin	1	75
8	(CH ₂) ₂ CH ₃	allyltributyltin	3	57
9	(CH ₂) ₃ CH ₃	tetrabutyltin	3	38
10	-	4-pyridinylboric acid ^e	4	86
11	Ph	phenylboric acid	2.5	92
12	-ОМе	4-methoxyphenylboric acid	2	80

a) For compounds 3-9 we used $(Ph_3P)_2PdCl_2$ as catalyst in HMPA at 65°C, for 10-12 we used $(Ph_3P)_4Pd$ in a mixture of toluene/ EtOH in the presence of Na_2CO_3 under reflux. b) Yields of isolated products calculated after coupling and hydrogenation reactions. c) For the synthesis of benzyltributyltin see reference 4b. d) The reaction time for hydrogenation was 4 hours instead of 18 hours. e) For the synthesis of 4-pyridinylboric acid see reference 7.

In this paper, we have reported a convenient preparation of 3-alkyl- or aryl-3,4-dihydro-2H-1-benzopyran derivatives 3 - 12⁸ using palladium catalysed reaction.

References and notes

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- 8. All new compounds have been fully characterised and their spectral data are in accord with the proposed structures.

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